

Available online at www.sciencedirect.com



Tetrahedron Letters

Tetrahedron Letters 48 (2007) 9171-9175

SmI₂-induced reductive cyclization of optically active β-alkoxyvinyl sulfoxides with aldehyde

Tomohiro Kimura, Mayumi Hagiwara and Tadashi Nakata*

Department of Chemistry, Faculty of Science, Tokyo University of Science, 1-3 Kagurazaka, Shinjuku-ku, Tokyo 162-8601, Japan

Received 19 September 2007; revised 16 October 2007; accepted 19 October 2007 Available online 24 October 2007

Dedicated to the memory of the late Professor Yoshihiko Ito

Abstract—SmI₂-induced reductive cyclization of optically active (*E*)- and (*Z*)- β -alkoxyvinyl sulfoxides with aldehyde was developed for the construction of several stereoisomers of tetrahydropyran derivatives. © 2007 Elsevier Ltd. All rights reserved.

Marine polycyclic ethers exemplified by brevetoxin-B, a red tide toxin, have a unique trans-fused polycyclic ether ring system.1 Their synthetically challenging complex structures and potent bioactivities have attracted the attention of numerous synthetic organic chemists. Thus, various methods for the synthesis of polycyclic ethers have been extensively studied.² We have already developed an efficient method for the construction of trans-fused polycyclic ether based on the SmI2-induced reductive cyclization of β -alkoxyacrylate A with a carbonyl group, affording 2,6-syn-2,3-trans-tetrahydropyrans **B** with complete stereoselectivity (Fig. 1).³ Several groups have successfully applied this method to the synthesis of polycyclic ethers.⁴ Recently, we have also reported SmI_2 -induced reductive cyclization of (E)and (Z)- β -alkoxyvinyl sulfones with aldehyde to give 2,6-syn-2,3-trans- and 2,6-syn-2,3-cis-tetrahydropyrans, respectively.⁵ We next turned our attention to the SmI₂-induced reaction of optically active β-alkoxyvinyl sulfoxides with aldehyde. The chirality of sulfoxide and the (E/Z)-stereochemistry of the olefin in the substrates would be expected to influence the stereoselectivity in these reactions. Lee and co-workers recently reported the same type of reaction using acyclic compounds.⁶ Here, we present our results on SmI₂-induced intramolecular cyclization of β-alkoxyvinyl sulfoxide with aldehyde.⁷



Figure 1. SmI₂-induced reductive cyclization.

First, we examined the SmI2-induced reductive cyclization with aldehydes **4b** and **6b** having (E)- β -alkoxyvinyl (R)- or (S)-sulfoxide, respectively, as substrates (Scheme 1). Addition of alcohol 1^{4f} to (R)-ethynyl *p*-tolylsulfoxide 2^8 in the presence of *N*-methylmorpholine (NMM) stereoselectively afforded (E)- β -alkoxyvinyl (R)-sulfoxide 4a in 96% yield,9 and this was reduced with DIBAH to give aldehyde **4b** in 85% yield. Treatment of (E)-(R)-**4b** with 2.5 equiv of SmI₂¹⁰ in the presence of MeOH (2.6 equiv) in THF effected reductive cyclization to give 2,6-anti-2,3-cis-tetrahydropyran 5a as a single product, which, without purification, was acetylated with Ac₂O to give acetate $5b^{11}$ in 64% yield (two steps). Use of CF₃CH₂OH instead of MeOH as a proton source slightly improved the yield of **5b** (71%).¹² On the other hand, the reaction of 1 and (S)-ethynyl p-tolylsulfoxide 3 in the presence of NMM, followed by DIBAH reduction, afforded aldehyde 6b. The SmI₂-induced cyclization of (E)-(S)-**6b** in the presence of MeOH afforded 2,6-syn-2,3-trans-tetrahydropyran 7a, which was acetylated to give acetate $7b^{11}$ in 85% yield (two steps).

Keywords: Samarium diiodide; C–C bond formation; Polycyclic ethers; Ethynyl *p*-tolylsulfoxide; Tetrahydropyranol.

^{*} Corresponding author. Tel.: +81 3 5228 8274; e-mail: nakata@ rs.kagu.tus.ac.jp

^{0040-4039/\$ -} see front matter @ 2007 Elsevier Ltd. All rights reserved. doi:10.1016/j.tetlet.2007.10.102



Scheme 1. Reagents and conditions: (a) (*R*)-2 or (*S*)-3, NMM, CH₂Cl₂, rt, 96% for 4a, 96% for 6a; (b) DIBAH, CH₂Cl₂, $-78 \degree C$, 85% for 4b, 92% for 6b; (c) SmI₂, MeOH, CH₂Cl₂, $0 \degree C$; Ac₂O, pyridine, rt, 64% for 5b (two steps), 85% for 7b (two steps); (d) SmI₂, CF₃CH₂OH, THF, $0 \degree C$; Ac₂O, pyridine, rt, 71% for 5b (two steps).



Scheme 2. Reagents and conditions: (a) LHMDS, (*R*)-2 or (*S*)-3, THF, 0 °C, then 1, -78 to -20 °C, 88% for 8a, 91% for 11a; (b) DIBAH, CH₂Cl₂, -78 °C, 45% for 8b, 73% for 11b; (c) SmI₂, MeOH, CH₂Cl₂, 0 °C; Ac₂O, pyridine, rt, 27% for 9b and 26% for 10b (two steps), 3% for 7b, 15% for 12b, and 21% for 13b (two steps); (d) SmI₂, CF₃CH₂OH, THF, 0 °C; Ac₂O, pyridine, rt, 45% for 9b and 27% for 10b (two steps).

Next, aldehydes **8b** and **11b**, having (Z)- β -alkoxyvinyl (R)- and (S)-sulfoxide, respectively, were examined (Scheme 2). Treatment of alcohol 1 and (R)-sulfoxide **2** with LHMDS stereoselectively afforded (Z)- β -alkoxyvinyl (R)-sulfoxide 8a in 88% yield,⁹ and DIBAH reduction gave aldehyde **8b** (45%). Treatment of (Z)-(R)-**8b** with SmI₂ in the presence of MeOH in THF followed by acetylation afforded two products; 2,6-syn-2,3-cis-tetrahydropyran **9b**¹¹ (27%) and γ -acetoxyvinyl sulfoxide 10b¹³ (26%). Use of CF₃CH₂OH instead of MeOH in the present reaction afforded 9b (45%) and 10b (27%). Moreover, addition of 1 and (S)-3 in the presence of LHMDS, followed by DIBAH reduction, afforded aldehyde 11b. The same reaction of (Z)-(S)-11b with SmI₂, followed by acetylation, gave many products, which contain 2,6-syn-2,3-trans-7b (3%), 2,6-syn-2,3-cis-12b¹¹ (15%), γ -acetoxyvinyl sulfoxide 13b¹³ (21%), etc. Use of CF₃CH₂OH did not improve the yield of 7b and 12b.

These results can be explained as follows (Fig. 2). In the SmI₂-induced cyclization, the first single electron reduction of aldehyde with SmI₂ gives a ketyl radical and then C–C bond formation occurs in the chelated intermediate to give the cyclized product.^{3,5} In the reaction of (E)-(R)-

4b with SmI₂, cyclization would proceed through transition state ii chelated by Sm(III) and sulfoxide to give 5a, because ii has an equatorial p-tolyl group in the chairlike conformation, whereas i has an axial one.¹⁴ Similarly, the reaction of (E)-(S)-**6b** would proceed through the chelated transition state iii having an equatorial *p*-tolyl group to give 7a. The reaction of (Z)-(R)-8b would also proceed through the chelated transition state v to give 9a. In the case of 11b, the corresponding chelated transition state vi would be unfavorable because of the axial *p*-tolyl group; thus, the reaction would proceed via the non-chelated transition state vii or viii to give 7a and 12a. The olefinic by-products 10a and 13a might be produced by ring opening subsequent to the cyclization; the axial-OSm(III) group of the intermediate ix, generated through v or viii via C-C bond formation followed by the second reduction with SmI₂, might participate in the ring opening together with the ring-O atom.

The *p*-tolylsulfoxymethyl group of product 7a was transformed to an aldehyde group for application to the synthesis of polycyclic ethers (Scheme 3). SmI₂-induced reaction of **6b** followed by TBS protection afforded the



Figure 2. Plausible transition states of SmI_2 -induced cyclization of 4b, 6b, 8b, and 11b.



Scheme 3. Reagents and conditions: (a) SmI₂, MeOH, THF, 0 °C; (b) TBSCl, imidazole, DMF, rt, 90% (two steps); (c) $(CF_3CO)_2O$, pyridine, MeCN, 0 °C; H₂O, K₂CO₃, 0 °C, 72%.



Scheme 4. Reagents and conditions: (a) SmI₂, MeOH, THF, 0 °C, 87% for 17a; (b) TBSCl, imidazole, DMF, rt, 88% for 17b; (c) (CF₃CO)₂O, pyridine, MeCN, 0 °C, then H₂O, K₂CO₃, NaBH₄, 97% for 18a; (d) *n*-Bu₄NF, THF, rt, 100% for 18b; (e) SmI₂, CF₃CH₂OH, THF, 0 °C, 68%.



Figure 3. Plausible transition state of SmI₂-induced cyclization of 17.

TBS-ether 14 in 90% yield. Sulfoxide 14 was subjected to the Pummerer rearrangement to give aldehyde 15 in 72% yield.

The stereospecific cyclization in the present reactions apparently proceeded via a chelated transition state involving strong coordination with sulfoxide and Sm(III). Therefore, we expected that reductive cyclization using acyclic aldehyde having an optically active β -alkoxyvinyl sulfoxide would be an effective approach for the asymmetric synthesis of tetrahydropyran derivatives. Treatment of aldehyde 16^{15} having (E)-(S)-vinylsulfoxide with SmI₂ in the presence of MeOH effected reductive cyclization to give the *trans*-tetrahydropyran 17a in 87% yield as a single product with >99% ee (Scheme 4).⁶ This result means that the reaction proceeds through the completely chelated transition state x (Fig. 3). Alcohol 17a was converted into (-)-3-tetrahydropyranol derivatives 18a¹⁶ and 18b¹⁶ in excellent vield via TBS protection, Pummerer rearrangement-NaBH₄ reduction, and removal of the TBS group. This method for the synthesis of (-)-18a and (-)-18b is expected to be useful, because these compounds were previously prepared from L-glucose.¹⁷ On the other hand, the reaction of aldehyde 19^{15} having (Z)-(S)-vinylsulfoxide with SmI₂ in the presence of CF₃CH₂OH gave cis-tetrahydropyran 20 in 68% yield. The same reaction using the corresponding enantiomers of 16 and 19 having (R)-vinylsulfoxide gave the enantiomers of 17a and 20, respectively.

In summary, the SmI₂-induced stereospecific cyclization of β -alkoxyvinyl sulfoxide with aldehyde was developed for the construction of several stereoisomers of tetrahydropyran derivatives. The desired stereoisomers of tetrahydropyrans could be obtained by selecting the appropriate combination of substrate and reagent, (R)-2 or (S)-3. Thus, asymmetric synthesis of 3-tetrahydropyranols was efficiently accomplished.

Acknowledgement

This work was financially supported by the Uehara Memorial Foundation and by a Grant-in-Aid for Scientific Research from the Ministry of Education, Culture, Sports, Science and Technology, Japan.

References and notes

- For reviews on polycyclic ethers, see: (a) Yasumoto, T.; Murata, M. Chem. Rev. **1993**, 93, 1897; (b) Shimizu, Y. Chem. Rev. **1993**, 93, 1685; (c) Murata, M.; Yasumoto, T. Nat. Prod. Rep. **2000**, 17, 293; (d) Yasumoto, T. Chem. Rec. **2001**, 1, 228; (e) Deranas, A. H.; Norte, M.; Fernández, J. J. Toxicon **2001**, 39, 1101.
- For reviews on synthetic methods and total syntheses, see:

 (a) Alvarez, E.; Candenas, M.-L.; Pérez, R.; Ravelo, J.; Martín, J. D. Chem. Rev. 1995, 95, 1953; (b) Fujiwara, K.; Hayashi, N.; Tokiwano, T.; Murai, A. Heterocycles 1999, 50, 561; (c) Mori, Y. Chem. Eur. J. 1997, 3, 849; (d) Marmsäter, F. P.; West, F. G. Chem. Eur. J. 2002, 8, 4347; (e) Inoue, M. Org. Biomol. Chem. 2004, 2, 1811; (f) Fujiwara, K.; Murai, A. Bull. Chem. Soc. Jpn. 2004, 77, 2129; (g) Sasaki, M.; Fuwa, H. Synlett 2004, 2, 1811; (h) Kadota, I.; Yamamoto, Y. Acc. Chem. Res. 2005, 38, 423; (i) Inoue, M. Chem. Rev. 2005, 105, 4379; (j) Nakata, T. Chem. Rev. 2005, 105, 4314; (k) Clark, J. S. Chem. Commun. 2006, 3571.
- (a) Hori, N.; Matsukura, H.; Matsuo, G.; Nakata, T. *Tetrahedron Lett.* **1999**, 40, 2811; (b) Hori, N.; Matsukura, H.; Nakata, T. Org. Lett. **1999**, 1, 1099; (c) Matsuo, G.; Hori, N.; Nakata, T. Tetrahedron Lett. **1999**, 40, 8859; (d)

Suzuki, K.; Matsukura, H.; Matsuo, G.; Koshino, H.; Nakata, T. *Tetrahedron Lett.* **2002**, *43*, 8653; (e) Matsuo, G.; Kadohama, H.; Nakata, T. *Chem. Lett.* **2002**, 148; (f) Hori, N.; Matsuo, G.; Matsukura, H.; Nakata, T. *Tetrahedron* **2002**, *58*, 1853.

- Selected papers (a) Sato, K.; Sasaki, M. Angew. Chem., Int. Ed. 2007, 46, 2518; (b) Fuwa, H.; Ebine, M.; Bourdelais, A. J.; Baden, D. G.; Sasaki, M. J. Am. Chem. Soc. 2006, 128, 16989; (c) Fuwa, H.; Kakinuma, N.; Tachibana, K.; Sasaki, M. J. Am. Chem. Soc. 2002, 124, 14983; (d) Kadota, I.; Takamura, H.; Sato, K.; Ohno, A.; Matsuda, K.; Satake, M.; Yamamoto, Y. J. Am. Chem. Soc. 2003, 125, 11893; (e) Nagumo, Y.; Oguri, H.; Shindo, Y.; Sasaki, S.; Oishi, T.; Hirama, M.; Tomioka, Y.; Mizugaki, M.; Tsumuraya, T. Bioorg. Med. Chem. Lett. 2001, 11, 2037; (f) Matsuo, G.; Kawamura, K.; Hori, N.; Matsukura, H.; Nakata, T. J. Am. Chem. Soc. 2004, 126, 14374; (g) Takahashi, S.; Kubota, A.; Nakata, T. Angew. Chem., Int. Ed. 2002, 41, 4751.
- 5. Kimura, T.; Nakata, T. Tetrahedron Lett. 2007, 48, 43.
- Jung, J. H.; Kim, Y. W.; Kim, M. A.; Choi, S. Y.; Chung, Y. K.; Kim, T.-R.; Shin, S.; Lee, E. Org. Lett. 2007, 9, 3225.
- (a) Kimura, T.; Nakata, T. Abstract of Papers, Part 2, p 1203, 87th Annual Meeting of the Chemical Society of Japan, Osaka, Japan, March 25–28, 2007; (b) Kimura, T.;

9.8, 4.6 Hz, 1H); 9b: δ 5.11 (broad, $W_{1/2} = 5.6$ Hz, 1H); 12b: δ 5.05 (broad, $W_{1/2} = 7.0$ Hz, 1H). NOEs between C2–H and C6–H in 9b and 12b were observed.

- Edmons, D. J.; Muir, K. W.; Procter, D. J. J. Org. Chem. 2003, 68, 3190.
- 13. Selected ¹H NMR data. 10b: δ 6.53 (dd, J = 15.3, 5.5 Hz, 1H), 6.38 (dd, J = 15.3, 1.2 Hz, 1H), 5.67 (m, 1H), 4.69 (ddd, J = 11.0, 9.8, 4.9 Hz, 1H); 13b: δ 6.49 (dd, J = 15.2, 5.8 Hz, 1H), 6.41 (d, J = 15.2 Hz, 1H), 5.65 (m, 1H), 4.66 (ddd, J = 10.9, 10.9, 4.8 Hz, 1H). Alkaline hydrolysis of the diacetate 10b followed by acetylation afforded a 2:3 mixture of 2,6-syn-2,3-cis-tetrahydropyran 9b and 2,6anti-2,3-trans-isomer via an intramolecular cyclization, and the same reaction of 13b predominantly afforded 2,6anti-2,3-trans-tetrahydropyran. These results confirmed the β-configuration of the 3-acetoxy group in 10b and 13b.
- 14. Lee et al. reported the same type of reaction using acyclic stereoisomers, including 21.⁶ They proposed similar transition states through sulfoxide and Sm(III) coordination to those shown here. However, they noted that it is difficult to propose a transition state structure for conversion of 21 to 22; a possible transition state structure xii does not adopt the familiar chair-like conformation. Their result would be well explained by our proposed transition state, i.e., the cyclization of 21 should proceed through the transition state xiii to give 22.



Hagiwara, M.; Nakata, T. Abstract of Papers, Part 4, p 20, 127th Annual Meeting of the Pharmaceutical Society of Japan, Toyama, Japan, March 28–30, 2007.

- (a) Kosugi, H.; Kitaoka, M.; Tagami, K.; Takahashi, A.; Uda, H. J. Org. Chem. 1987, 52, 1078; (b) Solladie, G.; Hutt, J.; Girardin, A. Synthesis 1987, 173.
- 9. Keum, G.; Kang, S. B.; Kim, Y.; Lee, E. Org. Lett. 2004, 6, 1895.
- (a) Namy, J. L.; Girard, P.; Kagan, H. B. Nouv. J. Chim. 1977, 1, 5; (b) Girard, P.; Namy, J. L.; Kagan, H. B. J. Am. Chem. Soc. 1980, 102, 2693; (c) Kagan, B. H. New J. Chem. 1990, 14, 453.
- 11. The stereochemistry of the newly formed tetrahydropyran in **5b**, **7b**, **9b** and **12b** was confirmed by the coupling constants of C3–H and NOE measurement. **5a**: δ 5.16 (ddd, J = 11.3, 5.5, 5.5 Hz, 1H); **7b**: δ 4.67 (ddd, J = 11.0,
- 15. Aldehydes 16 and 19 were prepared from 2,3-dihydrofuran as follows. (1) $BF_3 \cdot Et_2O$, MeOH, 1,3-propanedithiol, CH_2Cl_2 , rt;¹⁸ (2a) (*S*)-3, NMM, CH_2Cl_2 , rt, 85% (two steps) or (2b) (*S*)-3, LHMDS, THF, -20 °C, 92% (two steps); (3) MeI, CaCO₃, MeCN, H₂O, 60 °C, 88% for 16, 79% for 19.
- Synthesis of (+)-enantiomers of 18a and 18b from tri-Oacetyl-D-glucal: Nicolaou, K. C.; Hwang, C.-K.; Marron, B. E.; DeFrees, S. A.; Couladouros, E. A.; Abe, Y.; Carrol, P. J.; Snyder, J. P. J. Am. Chem. Soc. 1990, 112, 3040.
- (a) Shull, B. K.; Wu, Z.; Koreeda, M. J. Carbohyd. Chem. 1996, 15, 955; (b) Matsuo, G.; Hinou, H.; Koshino, H.; Suenaga, T.; Nakata, T. Tetrahedron Lett. 2000, 41, 903.
- 18. Köhnert, S. M.; Maier, M. E. Org. Lett. 2002, 4, 643.